

Stable Silylmethyl and Neopentyl Complexes of Scandium(III) and Yttrium(III)

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Summary The complexes $(\text{Me}_3\text{M}'\text{CH}_2)_3\text{M}\cdot 2\text{THF}$ ($\text{M} = \text{Sc}$ or Y ; $\text{M}' = \text{C}$ or Si) and $(o\text{-MeOC}_6\text{H}_4\text{SiMe}_2\text{CH}_2)_3\text{Sc}$ have been isolated from the reaction of the appropriate lithium reagent with MCl_3 or $\text{MCl}_3\cdot n\text{THF}$.

The n.m.r. data are consistent with a trigonal bipyramidal arrangement of ligands around the metal, the THF molecules occupying axial sites. The unusual five-co-ordination may be favoured by the rather bulky $\text{Me}_3\text{SiCH}_2^-$ ligands.

TABLE. Scandium and yttrium alkyls

Compound	M.p.	$\nu(\text{MC}_3)$ cm^{-1}a	N.m.r. (τ) ^b CH_2	Me
$(\text{Me}_3\text{CCH}_2)_3\text{Sc}\cdot 2\text{THF}^c$	66—68°	515	9.38	8.63
$(\text{Me}_3\text{SiCH}_2)_3\text{Sc}\cdot 2\text{THF}^c$	62—63°	450	10.27	9.72
$(\text{Me}_3\text{CCH}_2)_3\text{Y}\cdot 2\text{THF}^c$	78—79°	490	10.07 ^d	8.68
$(\text{Me}_3\text{SiCH}_2)_3\text{Y}\cdot 2\text{THF}^c$	48—50°	410	10.65 ^d	9.65
$(o\text{-MeOC}_6\text{H}_4\text{SiMe}_2\text{CH}_2)_3\text{Sc}$	115—120° (decomp.)	480	9.92	9.45

THERE are no well-defined binary alkyls of the lanthanides.¹ The reactions of the Group IIIA trichlorides, MCl_3 ($\text{M} = \text{Sc}$, Y , La), with methyl-lithium have been investigated but the products $[\text{Me}_3\text{M}]$, could not be freed from lithium or chloride impurities.² A sterically-crowded tetra-aryl-lanthanide complex $[\text{Li}(\text{THF})_4][\text{Lu}(\text{C}_6\text{H}_3\text{Me}_2-2,6)_4]$, has recently been characterised.³

We have prepared trimethylsilylmethyl and neopentyl complexes of scandium and yttrium, $(\text{Me}_3\text{M}'\text{CH}_2)_3\text{M}\cdot 2\text{THF}$ ($\text{M} = \text{Sc}$ or Y ; $\text{M}' = \text{C}$ or Si) by reaction of the appropriate lithium reagent with either the anhydrous metal chloride in hexane-ether-tetrahydrofuran (THF), or $\text{MCl}_3\cdot n\text{THF}$ in hexane-ether mixtures at ca. 0°. The complexes were obtained as analytically-pure, air-sensitive, colourless, crystals (*X*-ray studies are in hand) from *n*-pentane (Table). In the ¹H n.m.r. spectra, the methylene hydrogens of the yttrium complexes show coupling with ⁸⁹Y; yttrium is a rare example of a monoisotopic element having nuclear spin $I = \frac{1}{2}$.

^a Broad bands [*cf.*, 470 cm^{-1} in $(\text{Me}_3\text{SiCH}_2)_4\text{Ti}$, M. R. Collier, M. F. Lappert, and R. Pearce, *J.C.S. Dalton*, 1973, in the press]; ^b in C_6H_6 (2.73 τ); ^c $\nu(\text{C}-\text{O}-\text{C})$, 1020—1025 cm^{-1} (sharp) ^d doublet, $J(^{89}\text{YCH}^1\text{H})$ 2.5 Hz.

The co-ordinated THF could not be removed *in vacuo*, but by the use of the more bulky, and possibly chelating, ligand $o\text{-MeOC}_6\text{H}_4\text{SiMe}_2\text{CH}_2^-$, we have obtained a THF-free complex $(o\text{-MeOC}_6\text{H}_4\text{SiMe}_2\text{CH}_2)_3\text{Sc}$ (Table).

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¹ H. Gyslin and M. Tsutsui, *Adv. Organometallic Chem.*, 1970, **9**, 361; R. G. Hayes and J. L. Thomas, *Organometallic Chem. Rev.*, 1971, **7A**, 361.

² F. A. Hart, A. G. Massey, and M. S. Saran, *J. Organometallic Chem.*, 1970, **21**, 147.

³ S. A. Cotton, F. A. Hart, M. B. Hursthouse, and A. J. Welch, *J.C.S. Chem. Comm.*, 1972, 1225.